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Hexafluoroiodium(VII) hexafluoroarsenate

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The structure of hexafluoroiodium(VII) hexafluoroarsenate, IF₆AsF₆, has been determined by X-ray diffraction using a single crystal grown from a saturated solution in anhydrous HF. IF₆AsF₆ crystallizes in the cubic space group $Pa\overline{3}$ with a simple NaCl-like ionic packing. The I and As atoms occupy the 4*a* and 4*b* Wyckoff positions, respectively, with $\overline{3}$ symmetry.

Comment

Until now, the single-crystal X-ray structure of IF₆Sb₂F₁₁ represented the only detailed crystallographic study of salts containing the IF_6^+ cation (Lehmann *et al.*, 2004). IF₆AsF₆ was first prepared by the reaction of IF₇ with AsF₅ (Seel & Detmer, 1958, 1959). Powder X-ray diffraction studies of IF₆AsF₆ were performed later (Beaton, 1966; Christe & Sawodny, 1967). We have determined the structure of IF_6AsF_6 by single-crystal diffraction techniques to obtain more precise structural data and to allow a comparison of the cation geometry with that previously found in $IF_6Sb_2F_{11}$.

 IF_6AsF_6 crystallizes in the cubic space group $Pa\overline{3}$, in agreement with the published data based on X-ray powder diffraction experiments. The structure consists of alternating IF_6^+ and AsF_6^- octahedra packed in an NaCl-like arrangement (Fig. 1). The closest As \cdots I distance is 4.705 (1) Å. The I and As atoms are both located at sites with $\overline{3}$ symmetry (the 4a) and 4b Wyckoff positions, respectively), resulting in six equal I-F distances and six equal As-F distances. The I-F bond lengths [1.7744 (17) Å] are in the same range as those in IF₆Sb₂F₁₁ (1.767–1.782 Å at 173 K; Lehmann *et al.*, 2004). The As-F bond lengths [1.7200 (17) Å] are very similar to those in CsAsF₆ (1.714 Å; Loss & Röhr, 1998). The neigbouring IF₆⁺ and AsF_6^- octahedra are mutually tilted, possibly because of packing effects. The geometry of the IF_6^+ cation in IF_6AsF_6 is almost identical to that in $IF_6Sb_2F_{11}$.

Bond-valence analysis of IF₆AsF₆ gives bond-valence sums of 6.972 v.u. (bond valence units) for I and 4.578 v.u. for As (Brese & O'Keeffe, 1991), with contributions of 1.162 v.u. per F1 atom for the former and 0.763 v.u. per F2 atom for the latter.



Figure 1

24 parameters

A view of the unit-cell contents of IF_6AsF_6 . IF_6^+ and AsF_6^- ions are shown as light- and dark-grey octahedra, respectively.

Experimental

IF₆AsF₆ was synthesized by the oxidation of iodium pentafluoride by $[Ag(HF)_n]^{2+}$, prepared by reaction of AgF₂ with AsF₅ in anhydrous HF: $2[Ag(HF)_n^{2+} (AsF_6)_2^{-}] + IF_5 \Rightarrow 2(Ag^+ AsF_6^{-}) + IF_6AsF_6^{+} + IF_6AsF_6^{-}]$ AsF₅. Colourless block-shaped crystals of the title compound were grown from a saturated solution in anhydrous HF. Raman spectra recorded on single crystals were in agreement with literature data for IF₆AsF₆ (Beaton, 1966; Christe & Sawodny, 1967).

Crystal data	
AsF ₁₂ I $M_r = 429.82$ Cubic, $Pa\bar{3}$ a = 9.409 (2) Å V = 832.9 (3) Å ³ Z = 4	$D_x = 3.428 \text{ Mg m}^{-3}$ Mo K\alpha radiation $\mu = 7.96 \text{ mm}^{-1}$ $T = 200 \text{ K}$ Chunk, colourless $0.3 \times 0.3 \times 0.2 \text{ mm}$
Data collection	
Rigaku Mercury CCD diffractometer ω scans Absorption correction: multi-scan (Blessing, 1995) $T_{\rm min} = 0.115, T_{\rm max} = 0.2$	3317 measured reflections 354 independent reflections 230 reflections with $I > 2\sigma(I)$ $R_{int} = 0.018$ $\theta_{max} = 29.1^{\circ}$
Refinement	
Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.021$ $wR(F^2) = 0.063$ S = 1.10 354 reflections	$w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0225P)^{2}]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$ $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 0.72 \text{ e} \text{ Å}^{-3}$ $\Delta\rho_{min} = -0.60 \text{ e} \text{ Å}^{-3}$

 $\Delta \rho_{\rm min} = -0.60 \text{ e } \text{\AA}^{-3}$ Extinction correction: SHELXL97 Extinction coefficient: 0.0023 (4)

inorganic compounds

Data collection: *CrystalClear* (Rigaku, 1999); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *PATTY* in *DIRDIF92* (Beurskens *et al.*, 1992); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *DIAMOND* (Version 3.1; Bergerhoff *et al.*, 1996); software used to prepare material for publication: *WinGX* (Version 1.70; Farrugia, 1999), *enCIFer* (Version 1.2; Allen *et al.*, 2004) and *TEXAN* for Windows (Molecular Structure Corporation, 1999).

Supplementary data for this paper are available from the IUCr electronic archives (Reference: BC3005). Services for accessing these data are described at the back of the journal.

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